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Article

Acetylenic Fatty Acids and Stilbene Glycosides Isolated from *Santalum yasi* Collected from the Fiji Islands

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Abstract

In our continuing search for new anticancer and/or antimicrobial compounds from natural products we screened for these activities in bark and leaf extracts of sandalwood plants collected from the Fiji Islands and found *Santalum yasi* to be the most active. Resulting chemical workup enabled the isolation and structural characterization of a new acetylenic acid, methyl (E)-octadec-6-en-8-ynoate (1), and an atropisomeric stilbene glycoside (4) (Yasibeneoside) together with six known compounds: 11,13-octadecadien-9-ynoic acid (2), methyl octadeca-9,11-diynoate (3), gaylussacin (5) chrysin-7-beta-monoglucoside (6), neoschaftoside (7), and chrysin-6-C-glucoside-8-C-arabinoside (8). Compound 1 (18:2 (6t, 8a) is an example of an Δ^6 , Δ^8 acetylenic system containing the *trans* double bond at C-6 and the triple bond at C-8 which is reported here for the first time. All molecular structure elucidations and dereplications were performed using spectroscopic techniques including 2D NMR and HRMS-MS/MS spectrometry. Methyl (E)-octadec-6-en-8-ynoate showed moderate

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activity with an IC₅₀ of 91.2 ug/mL against the cancer human breast adenocarcinoma cell line MCF-7.

Keywords: Santalum yasi, Santalum album, Santalum yasi-album hybrid, acetylenic fatty acids Stilbene, cytotoxic

1. Introduction

Sandalwood represents a group of important medicinal and commercial plants belonging to the family Santalaceae, and the genus Santalum[1,2]. There are about 19 species of sandalwood known[3] of which one, S. yasi is endemic to Fiji[4,5] and Tonga[6]. A successful genetic hybrid between *S. yasi* and *S. album* (native to India)[7] is also growing in Fiji[8]. *San*talum yasi like other sandalwood is a hemi-parasitic plant with a very slow growth rate (about 0.3-0.7 m per year) with mature trees reaching heights of about 10 metres[4]. S. yasi has traditionally been used in Fiji for medicine, incense, and in wedding ceremonies[9]. Commercially, S. yasi has often been regarded to be of high value due to its high content in α -, and β -santalols that typically meets the East Indian Sandalwood ISO standard for sandalwood oil[5]. Most chemical investigative studies have been carried out on essential oil content of sandalwood[10-14], but studies on other compounds[14] in sandalwood such as S. yasi is lacking. To date no literature exists on the phytochemical content of leaves or bark of S. yasi making this study the first of its kind to shed light on the phytochemistry of S. yasi. While screening for cytotoxicity and antimicrobial activities in sandalwood plants from Fiji, significant activities were observed in the bark fractions of S. yasi, and the S. yasi-album hybrid species against methicillin-sensitive Staphylococcus aureus (MSSA ATCC-29213), and Candida albicans (C. albicans ATTCC-64124), and the cancer cell lines A549 (human lung carcinoma), A2058 (human Caucasian metastatic melanoma), HepG2 (human liver carcinoma cells), MCF-7 (human breast adenocarcinoma), MIA Paca-2 (human Caucasian pancreatic carcinoma), and PC-3 (human Caucasian prostate adenocarcinoma). Cytotoxicity activities were also observed in S. album leaf fractions against the cancer cell lines A2058, HepG2, and MCF-7. The report describes biological activities of fractions of the three sandalwood species in Fiji and the structure elucidation of compounds isolated from *S. yasi*.

2. Results and Discussion

Bark and leaves of the three sandalwood plants were collected in Fiji and extracted with methanol (MeOH) followed by dichloromethane (DCM) to produce six crude extracts. Each crude extract was then processed using a modified Kupchan[15] liquid-liquid partitioning method to produce four fractions based on polarity: water-butanol (WB), water-methanol (FM), dichloromethane (FD), and hexane (FH) (Figure S72). All fractions were screened for bioactivity against five clinically important human microbial pathogens (Table S70), and against six cancer cell lines (Table S71) with both *S. yasi* and the *S. yasi-album* hybrid bark fractions showing the most promising results (Figure 4). The *S. yasi* bark hexane fraction (SyB-FH) that had displayed good activity against MSSA and against four of the cancer cell lines was purified further on a Sephadex LH-20[16] column using MeOH-DCM (1:1) followed by reversed phase purification on a C₁₈ HPLC column using a H₂O-CH₃CN (40-60) solvent isocratic system to afford the new compound methyl (*E*)-octadec-6-en-8-ynoate (1), and the known compounds 11,13-octadecadien-9-ynoic acid (2), methyl octadeca-9,11-diynoate (3).

The dichloromethane fraction of *S. yasi* leaf (SyL-FD) was fractionated on a C₁₈ solid phase extraction column (SPE) using 50% H₂O-MeOH and further purified on a C₁₈ HPLC column using H₂O-MeOH gradient system to afford 2.8 mg of yasibeneoside (4), and the known compounds gaylussacin (5) chrysin-7-beta-monoglucoside (6), neoschaftoside (7), and chrysin-6-C-glucoside-8-C-arabinoside (8).

2.1 Structure Elucidation

Compound 1 showed a HRESIMS ion at m/z 293.2475 [M + H]⁺ (Δ 1.0 ppm) (Figure S11) for the expected molecular formula of C₁₉H₃₂O₂ with 4 indices of hydrogen deficiency[17,18]. Interpretation of ¹³C, HSQC, and HMBC NMR data (Figures S1-10), Table 1) of 1 showed the presence of 19 carbons in the

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form of 2 methyl groups (δc, 50.7, 13.0), 12 sp³ methylenes (δc 33.4, 32.5, 31.4, 28.7, 28.6, 28.5, 28.4, 28.3, 26.1, 24.6, 22.3, 18.5), 2 sp² methines (δc 142.5, 109.8), 2 non-protonated sp carbons (δc 87.7, 78.9), and one ester carbonyl (δc 174.8). Two sp² plus two sp carbons, plus one ester carbonyl accounted for the 4 indices of deficiency indicating the structure of 1 was linear. Interpretation of 1D and 2D NMR (Figures S1-10) data enabled the construction of two substructures (Figure 1). The position of the ester group was established by HMBC correlations of the methoxy proton at δ_H 3.60 (H-19), and methylenes at δ_H 2.27 (H-2), 1.56 (H-3) to the ester carbon at δ_C 174.8 (Table 1). Key COSY correlations between the methine proton at δ_H 5.92 (H-6), and δ_H 5.38 (H-7) to the methylene proton δ_H 2.02 (H-5) (Figure S4-5), HMBC correlations between the methine proton δ_H 5.92 (H-6) to the carbon at δ_C 32.5 (C-5) and the methylene protons at δ_H 2.27 (H-2), 1.56 (H-3), 2.02 (H-5) to the carbon at δ_C 28.6 (C-4) (Figures S7-9) unambiguously established the position of the alkene unsaturation at C-6/C-7. HMBC correlations between δ_H 5.38 (H-7) to the carbon at δ_C 87.7 (C-9) and between δн 5.92 (H-6) to δc 87.7 (C-8) (Figure S10, Table 1) established the 'ene-yne' spin system. The E geometry of the double bond in the molecule was secured through a ¹H-¹H coupling constant of 15.7 Hz between H-6 and H-7. Further evidence for the proposed structure came from NMR chemical shift predictions. A plot of ¹³C experimental data against predicted data calculated by ACD/Labs Structure Elucidator[19] using the HOSE[20] code is shown in Figure 2 with a linear regression of $r^2 = 0.9997$ suggesting that the proposed structure is most likely correct[21,22]. This strategy has been shown to be effective in predicting correct structures of natural products [22,23]. Additional evidence for the structure of 1 was provided by HR-MS/MS 99 100

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fragmentation data which has been annotated with fragments calculated by the ACD/labs MS Fragmenter[24] software (Figure S12). The presence of compound **1** as a methyl ester rather than an acid can be evidenced by the presence of proton NMR signals between δ_H 3-6-3-9 ppm on the semi-crude extract (1 H NMR not shown). In addition, the solvent system used for the extraction of **1** from the bark of *S. yasi* was also used to extract the seed oil of *S. album*. 1 H NMR of the seed oil of *S. album* showed the presence of ximenynic acid (Figure S74) and not its methyl ester. Both data suggest that compound **1** is a natural product rather than an artifact of extraction and purification.

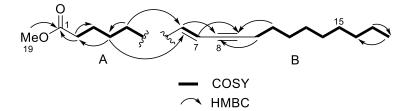


Figure 1. Substructures A and B showing key COSY and HMBC correlations establishing positions of unsaturation in the structure of 1.

Table 1. NMR Spectroscopic Data (600/150 MHz, CD₃OD) for Compound 1

			COSY	НМВС
Pos.	$\delta_{\rm C}$, type	$\delta_{\rm H}$ (<i>J</i> in Hz)	¹ H– ¹ H	¹ H→ ¹³ C
1	174.8, C			
2	33.4, CH ₂	2.27, t, 7.6	3	1, 3, 4
3	24.6, CH ₂	1.56, t, 7.4	2, 4	1, 2, 4, 6
4	28.6, CH ₂	1.28, ovlp ^a	3,4	6
5	32.5, CH ₂	2.02, m	4, 6, 7	6, 5
6	142.5, CH	5.92, m	4,7	8
7	109.8, CH	5.38, d, 15.7	5, 6, 5	9
8	78.9, C			

9	87.7, C			
10	18.5, CH ₂	2.20, t, 7.5	11	8, 9
11	28.5, CH ₂	1.45, ovlp	10	9
12	28.4, CH ₂	1.36, ovlp	11, 13	10
13	28.3, CH ₂	1.32, ovlp	12, 14	
14	28.2, CH ₂	1.28, ovlp	13,15	
15	28.1, CH ₂	1.29, ovlp	14,16	
16	31.4, CH ₂	1.24, ovlp	15, 17	15, 17
17	22.3, CH ₂	1.26, ovlp	16,17	18
18	13.0, CH ₃	0.85, t, 7.02	17, 19	16, 17, 31
19	50.7, CH ₃	3.60, s	18	1

 a Ovlp = overlap

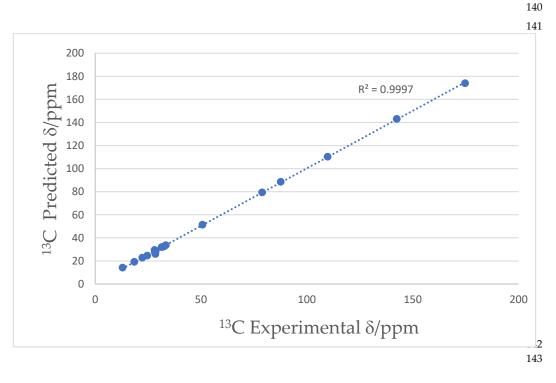


Figure 2. Plot of ¹³C experimental vs predicted chemical shifts data for 1.

Compound 4 showed a HRESIMS ion at m/z 837.2587 [M+H]⁺ calculated for C₄₂H₄₅O₁₈ (837.2606, Δ = 1.6 ppm) with 21 indices of hydrogen deficiency (Figure S27). Interpretation of Edited-HSQC, and HMBC NMR data (Figures S17-18, Table 2) indicated the presence of 42 carbons including 15 sp² methines (δ c 130.8, 129.0, 129.0, 128.9, 128.1, 127.4, 127.3, 126.3, 126.3, 125.4, 125.4, 111.0, 107.4, 102.9, 101.8), 10 sp³ methines (δ c 100.1, 99.9, 77.0, 76.9, 76.8, 76.5, 73.4, 69.9, 69.8), 4 sp³ methylenes (δ c 61.1, 61.0, 38.5, 37.9) and 13 sp² quaternary carbons (δ c 172.9, 172.7, 164.7, 164.0, 161.6, 161.5, 151.5, 147.1, 142.8, 142.0, 137.6, 106.4, 106.2). The 4 benzene ring systems plus one alkene plus one ester linkage plus one carboxylic acid group fully accounted for the 21 indices of hydrogen deficiency in the structure of 4. Extensive interpretation of one- and two-dimensional NMR data (Figures S15-18) enabled the construction of 6 substructures (Figure 3).

Figure 3. Substructures A-F derived from COSY (bold line) and HMBC correlations (H to C, blue arrows) for **4**.

The six substructures or spin systems include: one mono-substituted benzene ring coupled to an alkene system (substructure A), two 1,2,3,5-tetra-substituted benzene rings (substructures B and E), one 1,2-di-substitued benzene ring linked to an ethyl moiety (substructure D), and two glycosidic spin systems (substructures C and F). Long range HMBC correlations were used to connect these substructures to attain the full structure of 4. HMBC correlations were observed between the proton signal at δ_H 6.91 (H-23) to the sp² carbon at δ_C 142.8 (C-20) linking substructures A to B. Similarly, HMBC correlations were observed between the diastereotopic protons at δ_H 3.20/3.17 (H-8) to the sp² quaternary carbon at δ_C 147.1 (C-7) linking substructures D and E. Despite the absence of HMBC correlations between substructures B and D, substructure B was most likely linked to substructure D at C-15 based on the chemical shift of this carbon (δ_C 151.5) suggesting that C-15 was linked to C-16 via an ester linkage. A HMBC correlation between the anomeric proton at δ_H 5.05 (H-1") and the sp² quaternary carbon at 161.6 ppm (C-22) linked one of the glycosides

(substructure C) to substructure B. The second glycoside (substructure F) was linked to substructure E through an observed HMBC correlation between the anomeric proton at δH 4.89 (H-1') to the sp² quaternary carbon at δc 161.5 (C-5) to complete the planar structure for 4. Additional proof for the structure of 4 was obtained by plotting predicted and experimental ¹³C NMR data (Figure S26). A strong correlation (R² = 0.9996) was obtained suggesting that the proposed planar structure for 4 is most likely correct[21,25]. Further evidence for the structure for 4 was provided by analysis of HRMS/MS fragmentation data (Figure S28).

The geometry of the alkene system at C-23/C-24 was determined to be E based on the large coupling constant of 16.0 Hz between H-23 and H-24. The relative configurations of the two glycoside units were determined to be β based on coupling constant of 7.4 Hz between H-1' and H-2', and 6.9 Hz for H-1" and H-2." Further evidence was shown by NOE correlations between H-1' and H-3'/H-4'; H-2" and H-3"/H-4" (Table 2, Figures S19-22). NMR data show that the protons of the CH2 group at C-8, as well as C-9 resonate at different chemical shifts (Figures S15-19) suggesting chirality associated with atropisomers [26,27]. The tentative room temperature conformation of the structure of 4 is supported by key NOE correlations between H-8A/B and H-9A/B to H-11 and H-6 (Figures S19, S23-24). Figure S25 shows the minimized energy structure calculated using Chem3D Ultra[28]. The structure shows key H-bonding between the carbonyl group C-1 (carboxylic acid), and the hydroxy group at C-4", hindering free rotation between C-8 and C-9 resulting in chemical and magnetic non-equivalence of the methylene protons at C-8/C-9[29–31]. Due to sample limitation optical rotation and circular dichroism (CD) spectra were not measured for compound 4.

Table 2. NMR Spectroscopic Data (800/200 MHz, CD₃OD) for Compound 4

			COSY	HMBC	NOESY
Pos	δc, atype	δн (J/Hz)	¹ H- ¹ H	$^{1}H\rightarrow^{13}C$	
1	172.9, C				
2	106.4, C				
3	164.7, C				
4	101.8, CH	6.50, d, 7.4		<mark>1, 3, 5</mark>	
5	161.5, C				
6	111.0, CH	6.46, d, 2.3		1, 5, 7	8A, 8B
7	147.1, C				
8	38.5, CH ₂	A: 3.20, m	9	2, 6, 7, <mark>10</mark>	<mark>6, 11</mark>
		B: 3.17, m	9	2, 7, <mark>10</mark>	<mark>6, 11</mark>
9	37.9, CH ₂	A: 2.89, m	8A, 8B	7, <mark>10</mark>	<mark>6, 11</mark>
		B: 2.85, m	8A, 8B	7, <mark>10</mark>	<mark>6, 11</mark>
10	142.0, C				
11	128.2, CH	7.22, d, 8.0	12	<mark>10</mark> , 12, 15	8A, 8B, 9A, 9B
12	125.6, CH	7.17, t, 7.7	11	13	
13	128.0, CH	7.26, t, 8.0	14	12, 15	

14	127.6, CH	7.17, d, 8.0	13	12, 15	
15	151.5, C				
16	172.7, C				
17	106.2, C				
18	164.0 C				
19	102.9, CH	6.60, d, 2.3	21	16, 17, 18, 21, 23	
20	142.8, C				
21	107.4, CH	6.92, d, 2.2	19	16, 17, 19, <mark>22,</mark> 23	
22	161.6, C				
23	128.9, CH	6.91, d, 16.0	24	19, 20	
24	130.8, CH	7.98, d, 16.0	23	20, 24	
25	137.6, C				
26	126.3, CH	7.52, d, 8.0	28	<mark>25</mark> , 27, 28	
27	129.0, CH	6.95, d, 8.0	28, 26	<mark>25</mark> , 28	
28	128.0, CH	7.35, d, 8.0	27, 26		
<mark>29</mark>	129.0, CH	6.95, d, 8.0	<mark>28, 30</mark>	<mark>25, 28</mark>	
<mark>30</mark>	126.3, CH	7.52, d, 8.0	<mark>28, 29</mark>	<mark>25, 27, 28</mark>	
1'	99.9, CH	4.74, d, 7.0	2'	2', <mark>5</mark>	3' 4'
2'	73.4 CH	3.49, m	1', 3'		
3'	76.8, CH	3.39, m	2', 4'	2'	<mark>1'</mark>
4'	69.8, CH	3.48, m	3'		<mark>1'</mark>
5'	76.5, CH	3.44, m	4'	5'	
(1	(1.0. CII	A: 3.91, dd,			
6'	61.0, CH ₂	11.8, 2.2	5'		
		B: 3.72, m	5'		
1"	100.1, CH	5.05, d, 7.0	2"	<mark>22</mark>	3'', 4''
2"	73.4, CH	3.50, m	1", 4"	2''	
3"	76.9, CH	3.46, m	2", 4"		<mark>1''</mark>
4''	69.9, CH	3.41, m	3"	2''	<mark>1''</mark>
5"	77.0, CH	3.53, m	4", 6"A, 6"B		
<i>(</i> !!	61.1, CH ₂	A: 3.93, dd,			
6"		11.8, 2.2	5''	5''	
		B: 3.72, m	5"		

^aCarbons extracted from 2D NMR (HSQC and HMBC)

Compound 2 showed an m/z of 291.2318 [M+H]⁺ calculated for C₁₉H₃₁O₂ (291.2319, Δ = -0.2

ppm) (Figure S40). Interpretation of 1 H, 2D NMR and HR-MS/MS data (Figures S31-39, S41-42) identified **2** as 11,13-octadecadien-9-ynoic acid[32]. Compound **3** showed an m/z of 291.2319 [M+H]+ calculated for $C_{19}H_{31}O_2$ (291.2319, Δ = 0.4 ppm) (Figure S48). Interpretation of 1 H, 2D NMR and HR-MS/MS data (Figures S43-47,

S49) identified compound 3 as methyl octadeca-9,11-diynoate[33].

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Compound 5 showed an m/z of 419.1347 M+H]⁺ calculated for C₂₁H₂₃O₉ (419.1342, Δ = -2.5 ppm) (Figure S54). Interpretation of ¹H, and 2D NMR data (Figures S50-53) identified compound 5 as gaylussacin[33,34].

Compound **6** showed an m/z of 417.1191 M+H]⁺ calculated for C₂₁H₂₁O₉ (417.1186, Δ = -2.6 ppm) (Figure S59). Interpretation of ¹H, ¹³C and 2D NMR data (Figures S55-58) identified the compound as chrysin-7-beta-monoglucoside[35,36].

Compound 7 showed an m/z of 565.1570 M+H]⁺ calculated for C₂₆H₂₉O₁₄ (565.1557, Δ = 2.9 ppm) (Figure S64). Interpretation of ¹H, and 2D NMR data (Figures S60-63) identified the compound as neoschaftoside[37,38].

Compound 8 showed an m/z of 549.1618 M+H]⁺ calculated for C₂₆H₂₉O₁₃ (549.1608, Δ = 2.9 ppm) (Figure S69). Interpretation of ¹H, and 2D NMR data (Figures S65-68) identified the compounds as chrysin-6-C-glucoside-8-C-arabinoside[39].

2.2 Biological Activity

The modified Kupchan fractions of leaf and bark of *S. yasi, S. album,* and *S. yasi-album* hybrid were screened for antibacterial, antifungal, and cytotoxic activities. Leaf fractions of the three sandalwood plants displayed low to mild activity at 0.32 mg/mL against the pathogens tested (Table S70) with the butanol (WB) fraction of the bark of *S. yasi* showed good activity against *Candida albicans*. The strongest activities were shown by the FH fractions of the bark of *S. yasi,* and *S. yasi-album* hybrid against MSSA ATCC-29213 (>90% inhibition) (Figure 4b, Table S70).

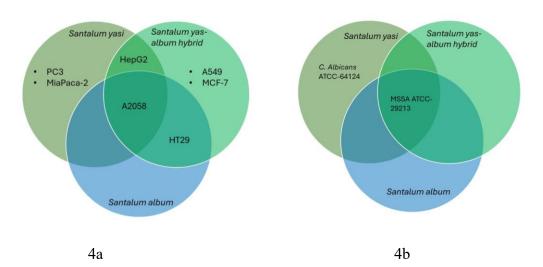


Figure 4a. Cytotoxic activity of the bark hexane fractions of *S. yasi, S. album*, and *S. yasi-album* hybrid. A549 (human lung carcinoma), A2058 (human Caucasian metastatic melanoma), HepG2 (human liver carcinoma cells), MCF7 (human breast adenocarcinoma), MiaPaca-2 (human Caucasian pancreatic carcinoma), and PC-3 (human Caucasian prostate adenocarcinoma), HT-29 (human colorectal adenocarcinoma). Both *S. yasi* and *S. yasi-album* hybrid were active at 0.10 mg/mL while *S. album* was active at 0.05 mg/mL (Table S71). 4b. Antimicrobial assay activity. MSSA ATCC-29213 (Methicillin-susceptible *Staphyloccocus aureus* subspecies *aureus* strain Wichita), *C. albicans* ATCC-64124 (*Candida albicans*). The *C. albicans* activity (>50% inhibition, Table S70) was reported only in the WB fraction while the MSSA activity

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were recorded in the bark FH fractions of the three sandalwood plants. Antimicrobial assays of fractions were performed at 0.06 mg/mL.

For the cytotoxicity assay both *S. yasi* bark (FH), and *S. album* leaf (FM) showed activity (60-90%) at 0.10 mg/mL against human Caucasian metastatic melanoma (A2058), human liver carcinoma cells (HepG2), human breast(adenocarcinoma) (MCF7), human Caucasian pancreatic carcinoma (MIA PaCa-2), and human Caucasian prostate adenocarcinoma (PC-3) with the strongest activity shown by the *S. yasi* bark fraction (FH) at 90% inhibition of HepG2 cells. *S. yasi-album* hybrid bark (FH) inhibited both human lung carcinoma (A549) and human colorectal adenocarcinoma (HT-29) in addition to the cell lines inhibited by *S. yasi and S. album* (Figure 4a, Table S71).

Methyl (*E*)-octadec-6-en-8-ynoate (**1**) showed an IC₅₀ of 91.2 ug/mL (Figure S73) but showed no activity against A2058 and human fibroblast cells (MRC-5). Compounds **5-8** showed weak cytotoxic activity (results not shown).

3. Materials and Methods

3.1. Reagents and Solvents

All solvents used for chromatography purifications were HPLC grade, while LCMS solvents were MS grade, both were obtained from Fisher Scientific[40], and NMR solvents were obtained from Goss Scientific[41].

3.2 Main Instruments

UV spectra were recorded on an Agilent Technologies 1220 Infinity Photodiode array detector[42]. IR spectra were recorded on a Shimadzu fourier transform infrared spectrophotometer (IRTracer-100)[43]. NMR spectroscopic data for compounds 4-8 were recorded at the University of Edinburgh[44] at 25 °C on a Bruker Avance NEO 800 MHz with a Hecooled cryoprobe Compounds 1-3 were recorded at the Biodiscovery Centre, University of Aberdeen[45] on a Bruker AVANCE III HD Prodigy TCI cryoprobe at 600 and 150 MHz for ¹H and ¹³C, respectively. This instrument was optimized for ¹H observation with pulsing/decoupling of ¹³C and ¹⁵N with ²H lock channels equipped with shielded z-gradients and cooled preamplifiers for ${}^{1}\!H$ and ${}^{13}\!C$. The ${}^{1}\!H$ and ${}^{13}\!C$ chemical shifts were referenced to the solvent signals (δ_H 3.31 and δ_C 49.00 in CD₃OD). LC-HRESIMS analysis for compounds 1-4 were performed on an Agilent 1290 LC system with a photodiode array detector (DAD) coupled to an Agilent 6546 LC-QTOF equipped with dual spray jet stream technology electrospray ion source (AJS)[42]. The system was controlled by the Mass Hunter 11.0[46] software. Liquid chromatographic separations were performed at 40 °C on a Kinetex phenyl hexyl 100 × 3.0 mm, 1.7 μm (Phenomenex)[47] equipped with a security guard column. A linear CH₃CN-H₂O gradient of 20% CH₃CN-water to 100% CH₃CN in 12 min was applied at a constant flow of 0.4 mL/min; then 100% CH₃CN was maintained for a 3 min before returning to the starting conditions in 1 min and equilibrating for a further 3 min. Formic acid (0.1% v/v) was added to all solvents and UV spectra were collected by a DAD from 200 to 500 nm with a resolution of 2 nm. Tuning in ESI* mode was performed using Agilent's tuning mix[42] of 6 masses to a resolution of FWHM between 40,495 for [M+H]+ of

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118.086255 at the lower end and 64,549 for 1521.971475 at the upper end giving rise to an average accuracy of <1.0 ppm within the mass range. Mass accuracy was maintained throughout sample analysis via use of dual spray technology using Agilent's reference solution mix[42]. Scanning source parameters for ESI $^+$ were as follows: capillary (3,500 V), nozzle (1,000 V), fragmentor (190 V), skimmer1 (65 V), and octapole RF peak (750 V). Targeted MS 2 (ES $^+$) fragmentation mode for compound **1** (m/z 293.2501) was performed at four collision energies: 10, 20, 30, 40 V at peak retention time (t_R) of 9.80 +/- 0.06 min. Targeted MS 2 (ES $^-$) fragmentation mode for compound **4** (m/z 835.2455) was performed at four collision energies: 5, 10, 15, 30 V at peak retention time (t_R) of 3.08 +/- 0.06 min.

LC-HRESIMS for compounds **5-8** were performed on a Bruker Maxis II Time of Flight instrument[45,48] using the following parameters: capillary voltage 45 V, capillary temperature 320 °C, auxiliary gas flow rate 10–20 arbitrary units, sheath gas flowrate 40–50 arbitrary units, spray voltage 4.5 kV, mass range 100–2000 amu, resolution 80 000 for HRESIMS.

3.3 Chomatography

Sephadex LH-20[49] was sourced from Merk[16]. Solid phase extractions were performed using Phenomenex C18 cartridges (Strata C18-E, 55 um, 70 Å)[50]. Semipreparative HPLC purifications were performed on an Agilent 75 1100 HPLC system consisting of a binary pump, degasser, photodiode array detector.

3.4 Plant Collection and Extraction

Sandalwood leaves and bark were collected from the Fiji Islands: S. yasi from 114 Milverton road, Suva (-18.1349, 178.4476)[51], S. album from Vavalagi road, Nakasi (-18.06517, 178.5154)[51,] and S. yasi-album hybrid from Yauvula, Wainunu, Bua (-16.805842, 178.889114)[51]. All samples were taken to the Institute of Applied Sciences, University of the South Pacific[52], Suva for processing. All plants were taxonomically identified based on morphological features at the South Pacific Regional Herbarium[53] where voucher specimens are kept with the following collection numbers: S. yasi (Tuiwawa5237), S. album (Tuiwawa5238), and <mark>S. yasi-album</mark> hybrid (Tuiwawa5239). Bark (10 g) and leaves (5 g) were extracted separately with methanol (3×) followed by dichlomethane (DCM 3×). Extracts were dried under vacuum, and fractionated using the modified Kupchan liquid-liquid partitioning technique[15,54](Figure S72) to produce four fractions: hexane (FH), dichloromethane (FD), methanol-water (FM) and butanol-water (WB). The process is summarized as follows: The crude sample was dissolved in water (30 mL) and partitioned with DCM (30 mL). The aqueous layer was then partitioned with secondary butanol (30 mL) to produce the WB fraction. The DCM fraction was dried under rotary evaporation, dissolved in 90% methanol-water (30 mL), and partitioned with n-hexane (30 mL) to yield the FH fraction. The 90% methanol-water layer was phase adjusted to 50% methanol-water and partitioned with DCM (54 mL) to yield the FM and FD fractions respectively. Each of the steps described above was repeated three times to enhance separation efficiencies. These four fractions (WB, FM, FD, and FH) were dried and shipped to the University of Lancashire in the United Kingdom.

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3.5 Purification and Isolation

The FH fraction of the bark of *S. yasi* was subjected to size exclusion chromatography using the Sephadex LH-20 gel as stationary phase eluted with dichloromethane-MeOH (1:1) to yield 10 Sephadex fractions. The most interesting fraction in terms of ¹H NMR profile was fraction Sephadex-4 (S-4) which was subjected to HPLC purification on a Waters Sunfire C18 OBD™ semi-Prep column (250×10 mm) using a solvent gradient from 0 to 100% 336 CH₃CN in 25 min and maintained at 100% CH₃CN for a further 10 min with a flow rate of 1.5 mL/min to yield 2.5 mg (0.025% ww) of 1, 2.3 mg (0.023% ww) of 2, and 1.8 mg (0.018% ww) of 3.

The FD fraction of the leaf extract of S. yasi was fractionated further into four sub-fractions using reversed phase C-18 SPE cartridges where after conditioning of the column as recommended by Phenomenex[50], the column was flushed with 100% water to remove salts and other polar compounds followed by 25%, 50% and 100% (MeOH-H₂O). After drying under nitrogen flow, fractions were profiled by 1H NMR with the SPE-50% fraction selected for further purification on an ACE 5 C18 HL 250 × 10 mm using an isocratic solvent system of 30% CH₃CN-H₂O at a flow rate of 1.5 mL/min to yield 3.0 mg (0.060% ww) of 4, 1.9 mg (0.038% ww) of **5**, 1.8 mg (0.036% ww) of **6**.

3.6 Cytotoxicity Assays

Cytotoxicity assays for compound 1 were performed as follows following the method described by Schlüter et. al.[55]: Human melanoma cells (A2058, ATCC no: CRL-1147) were grown and assayed in Dulbeccos Modified Eagles Medium (DMEM, Sigma D6171) supplemented with fetal bovine serum (FBS), glutamine stable and gentamycin. Human breast adenocarcinoma (MCF7, ATCC no HTB-22) and human lung fibroblasts, normal (MRC-5, ATCC no: CCL-171) were grown and assayed in minimum essential medium eagle (MEM, Sigma M7278) supplemented with FBS, glutamine stable, non-essential amino Acids, sodium pyruvate and gentamycin. Cells were seeded in 96-well microtiter plates and incubated for 24 hours at 37°C and 5 % CO2. After 24 hours the samples were added at a concentration of 100 µg/mL, and the cells were incubated for 72 hours at 37°C and 5 % CO₂. Cell seeding density were 2000 cells/well for A2058 and MCF7 respectively. MR-C5 cells were seeded at 4000 cells/well. Cytotoxicity assay for 1 were measured by adding CellTiter 96 Aqueous One Solution (Promega). Metabolically active cells will reduce the yellow MTS salt to purple formazan. After 60 minutes of incubation with Aqueous One Solution, absorbance was read at 490 nm. The quantity of formazan measured at 490 nm is directly proportional to the number of living cells. Cell survival is then calculated by comparing the samples with a negative control (100 % cell survival) and a positive control (100 % cell death).

The cytotoxic activity of sandalwood fractions were tested against seven different human cancer cell lines: A549 (lung carcinoma), A2058 (metastatic melanoma), MCF7 (breast adenocarcinoma), HT-29 (colorectal adenocarcinoma MIA PaCa-2 (pancreatic carcinoma), PC-3 (prostate adenocarcinoma), and HepG2 (hepatocyte carcinoma) based on the MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay[56]. Fractions were tested in duplicate following an established methodology[57].

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3.7 Antimicrobial Assays

Antimicrobial testing against human pathogenic Gram Positive (methicillin sensitive *Staphylococcus aureus* (MSSA) ATCC29213 and methicillin resistant *Staphylococcus aureus* (MRSA) MB5393) and Gram Negative bacteria (*Escherichia coli* ATCC 25922, *Klebsiella pneumoniae* ATCC700603, *Pseudomonas aeruginosa* PAO-1 and *Acinetobacter baumannii* MB5973), yeast (*Candida albicans* ATCC64124) and fungi (*Aspergillus fumigatus* ATCC46645) were performed following established procedures[58]. Fractions were tested in duplicate.

4. Conclusions

The study has shown the potential of sandalwood plants as source of cytotoxic compounds in particular S. yasi (and S. yasi-album hybrid) resulting in the isolation and characterization of two new compounds, one of which compound 1, an acetylenic acid, showing moderate cytotoxic activity against breast cancer cells, MCF-7. Even though acetylenic acids (1-3) have been isolated for the first time from the bark of S. yasi, this class of compound has been previously identified in sandalwood. Ximenynic acid has been known to occur in seeds of S. album, S. insulare, and others[13,59] and the compound is known for its various biological activities including anti-inflammatory, anticancer, larvicidal, and antimicrobial, and widely used in the cosmetic industries[60,61]. Unsaturated fatty acid with double and/or triple bonds have been of interest due to its potency against fungal pathogens[62] with undecylenic acid (UDA) is an example of such compound is on the market as an antifungal agent[63]. However, the position of the double and triple bond can vary[64] and bioactivity is dependent on the length of the fatty acid chain and position of unsaturation[63]. Known acetylenic acids previously isolated from Santalum sp. have the usual Δ^9 , Δ^{12} unsaturation carbons[64]. Compound 1 has the *trans* double bond at C-6 and a triple bond at C-8 which is different from those previously reported in literature where C-6 is either a cis-double bond conjugated to cis double bonds at C-9[65] or a triple bond at C-6 either alone or in conjugation with cis/trans double[65,66] or triple bonds[63]. The closest related compounds are 6-octadecen-9-ynoic acid isolated from the nuts of Ongokea klaineana [67] which has a trans double bond at C-6, and a triple bond at C-9, and 9-octadecen-6-ynoic acid isolated from Riccia fluitans [66] where the trans double bond is at C-9, and the triple bond at C-6, hence, compound 1 is the first example of an unusual C-6/C-8 unsaturations. Furthermore, alkyne bond containing natural products such as polyacetylenes have been known for their strong antimicrobial activities[68,69] against drug resistant strains such as MRSA, [70,71] and anticancer activities [72] making them potential drug lead templates. Yasibeneoside (4) is the first example of a stilbene in the family Santalaceae (40 genera)[73] as stilbenes have been known only to occur in other plant families such as Vitaceae, Leguminaceae, Gnetaceae, and Dipterocaraceae[74] indicating the wide occurrence of the enzyme stilbene synthase (STS)[75] in other plant families. It adds a new structure to the stilbene family of structures that have been extensively studied for their cancer preventative, and tumour suppression effects[76-80]. Stilbene containing scaffolds such as tamoxifen and raloxifene are synthetic approved FDA drugs in clinical and have been known to lower risk for breast cancer[81] underpinning the importance of this scaffold in drug discovery and development. In addition, yasibeneoside displays atropisomerism, a property that is attracting a lot of interest in the drug discovery and drug

development field due to potential effects of various conformations on receptor binding in	418
biological systems[27,82]. However, due to sample limitation in this study, bioactivity in-	419
vestigation for yasibeneoside was not carried out.	420
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Supplementary Materials: The following supporting information can be downloaded at:	422
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